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11) Which molecule has a dipole moment of zero? A) CHCl_3 B) CH_2Cl_2 C) $\text{ClHC}=\text{CH}_2$ D) *trans*- $\text{ClHC}=\text{CHCl}$ E)

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None of these choices. Answer: D Topic:

Molecular geometry, polarity Section

Reference 1: 1.5 and 2.3 Difficulty:

Medium 12) Which molecule would have
a dipole moment greater than zero? A)

BeCl₂ B) BCl₃ C) CO₂ D) H₂O E) CCl₄

Answer: D

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